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- Q 1. A material whose K absorption edge is  $0.15 \text{ \AA}$  is irradiated with  $0.1 \text{ \AA}$  X-rays. The maximum kinetic energy of photoelectrons that are emitted from K-shell is –  
(a) 41 KeV                      (b) 51 KeV                      (c) 61 KeV                      (d) 71 KeV
- Q 2. For a given material, the energy and wavelength of characteristic x-ray satisfy–  
(a)  $E(K\alpha) > E(K\beta) > E(K\gamma)$   
(b)  $E(M\alpha) > E(L\alpha) > E(K\alpha)$   
(c)  $\lambda(K\alpha) > \lambda(K\beta) > \lambda(K\gamma)$   
(d)  $\lambda(M\alpha) > \lambda(L\alpha) > \lambda(K\alpha)$

### COMPREHENSION (Q3 to Q5)

A tungsten target ( $z = 74$ ) is bombarded by electrons in an X-ray tube. The K, L and M atomic X-rays energy levels for tungsten are 69.5, 11.3 and 2.3 keV

- Q 3. The minimum value of the accelerating potential that will permit the production of the characteristic  $k\beta$  and  $k\alpha$  lines of tungsten–  
(a) 69.5 kV                      (b) 58.2 kV                      (c) 67.2 kV                      (d) 11.3 Kv
- Q 4. For the same accelerating potential, what is  $\lambda_{min}$ ?  
(a) 16.9 pm                      (b) 17.9 pm                      (c) 18.9 pm                      (d) 19.9 pm
- Q 5. What is  $k\beta$  wavelength ?  
(a) 16.5 pm                      (b) 17.5 pm                      (c) 18.5 pm                      (d) 21.5 pm
- Q 6.  $K\alpha$  wavelength emitted by an atom of atomic number  $Z = 11$  is  $\lambda$ . Find the atomic number for an atom that emits  $K\alpha$  radiation with wavelength  $4\lambda$  :  
(a)  $Z = 6$                       (b)  $Z = 4$                       (c)  $Z = 11$                       (d)  $Z = 44$
- Q 7. X-rays will not show the phenomenon of :  
(a) diffraction                      (b) polarisation  
(c) deflection by electric field                      (d) interference
- Q 8. The wavelength of  $K\alpha$  X-rays produced by an X-ray tube is  $0.76 \text{ \AA}$ . The atomic number of the anticathode material is  
(a) 82                      (b) 41                      (c) 20                      (d) 10



- Q 9. The minimum wavelength of X-ray that can be produced in a Coolidge tube depends on
- the metal used as the target
  - the intensity of the electron beam striking the target
  - the current flowing through the filament
  - the potential difference between the cathode and the anode
- Q 10. If the potential difference applied across a Coolidge tube is increased,
- the wavelength of the  $K_{\alpha}$  line will increase
  - the wavelength of the  $K_{\beta}$  line will decrease
  - the difference in wavelength between the  $K_{\alpha}$  and  $K_{\beta}$  lines will decrease
  - none of the above
- Q 11. If the voltage applied to a X-ray tube is being increased  $\eta = 1.5$  times, the short wave limit of the X-ray continuous spectrum shifts by  $\Delta\lambda = 26$  pm. Find the initial voltage applied to the tube.
- 10 KV
  - 12 KV
  - 14 KV
  - 16 KV
- Q 12. The potential difference applied to an X-ray tube is 5 kV and the current through it is 3.2 mA. Then the number of electrons striking the target per second is
- $2 \times 10^{16}$
  - $5 \times 10^6$
  - $1 \times 10^{17}$
  - $4 \times 10^{15}$
- Q 13. When the voltage applied to an X-ray tube is increased from  $V_1 = 10$  kV to  $V_2 = 20$  kV, the wavelength interval between the  $K_{\alpha}$  line and the short-wave cut off of the continuous X-ray spectrum increases by a factor  $\eta = 3.0$ . Find the atomic number of the element of which the tube's anticathode is made.
- 11
  - 20
  - 29
  - 38
- Q 14. If a potential difference of 20,000 volts is applied across an X-ray tube, the cut-off wavelength will be
- $6.21 \times 10^{-10}$  m
  - $6.21 \times 10^{-11}$  m
  - $6.21 \times 10^{-12}$  m
  - $3.1 \times 10^{-11}$  m
- Q 15. Which of the following pairs constitute very similar radiations?
- Hard ultraviolet rays and soft X-rays
  - Soft ultraviolet rays and hard X-rays
  - Very hard X-rays and low-frequency  $\gamma$ -rays
  - Soft X-rays and  $\gamma$ -rays
- Q 16. When an electron moving at a high speed strikes a metal surface, which of the following are possible?
- The entire energy of the electron may be converted into an X-ray photon.



- (b) Any fraction of the energy of the electron may be converted into an X-ray photon.
- (c) The entire energy of the electron may get converted to heat.
- (d) The electron may undergo elastic collision with the metal surface

- Q 17. The wavelength of  $K_{\alpha}$  X-rays for lead isotopes  $Pb^{208}$ ,  $Pb^{206}$ ,  $Pb^{204}$  are  $\lambda_1$ ,  $\lambda_2$  and  $\lambda_3$  respectively. Then :
- (a)  $\lambda_1 = \lambda_2 = \lambda_3$
  - (b)  $\lambda_1 > \lambda_2 > \lambda_3$
  - (c)  $\lambda_1 < \lambda_2 < \lambda_3$
  - (d)  $\lambda_2 = \sqrt{\lambda_1 \lambda_3}$

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## Answer Key

Q.1 a	Q.2 c,d	Q.3 a	Q.4 b	Q.5 c
Q.6 a	Q.7 c	Q.8 b	Q.9 d	Q.10 d
Q.11 d	Q.12 a	Q.13 c	Q.14 b	Q.15 a,c
Q.16 a,b,c	Q.17 a,d			


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Physics DPP - Solution**

**DPP: X- Ray**

**By Physicsaholics Team**

Q1) A material whose K absorption edge is  $0.15 \text{ \AA}$  is irradiated with  $0.1 \text{ \AA}$  X-rays. The maximum kinetic energy of photoelectrons that are emitted from K-shell is -

$$\text{Energy required to remove K shell electron} = \frac{hc}{\lambda} = \frac{12400 \text{ eV}}{0.15 \text{ \AA}}$$

$$\begin{aligned} \text{Energy of photon} &= \frac{12400 \text{ eV}}{0.1 \text{ \AA}} = 124 \text{ KeV} \\ &= \frac{124 \text{ KeV}}{0.15} \\ &= \frac{1240}{15} \text{ KeV} = 82.7 \text{ KeV} \end{aligned}$$

(a) 41 KeV

(b) 51 KeV

(c) 61 KeV

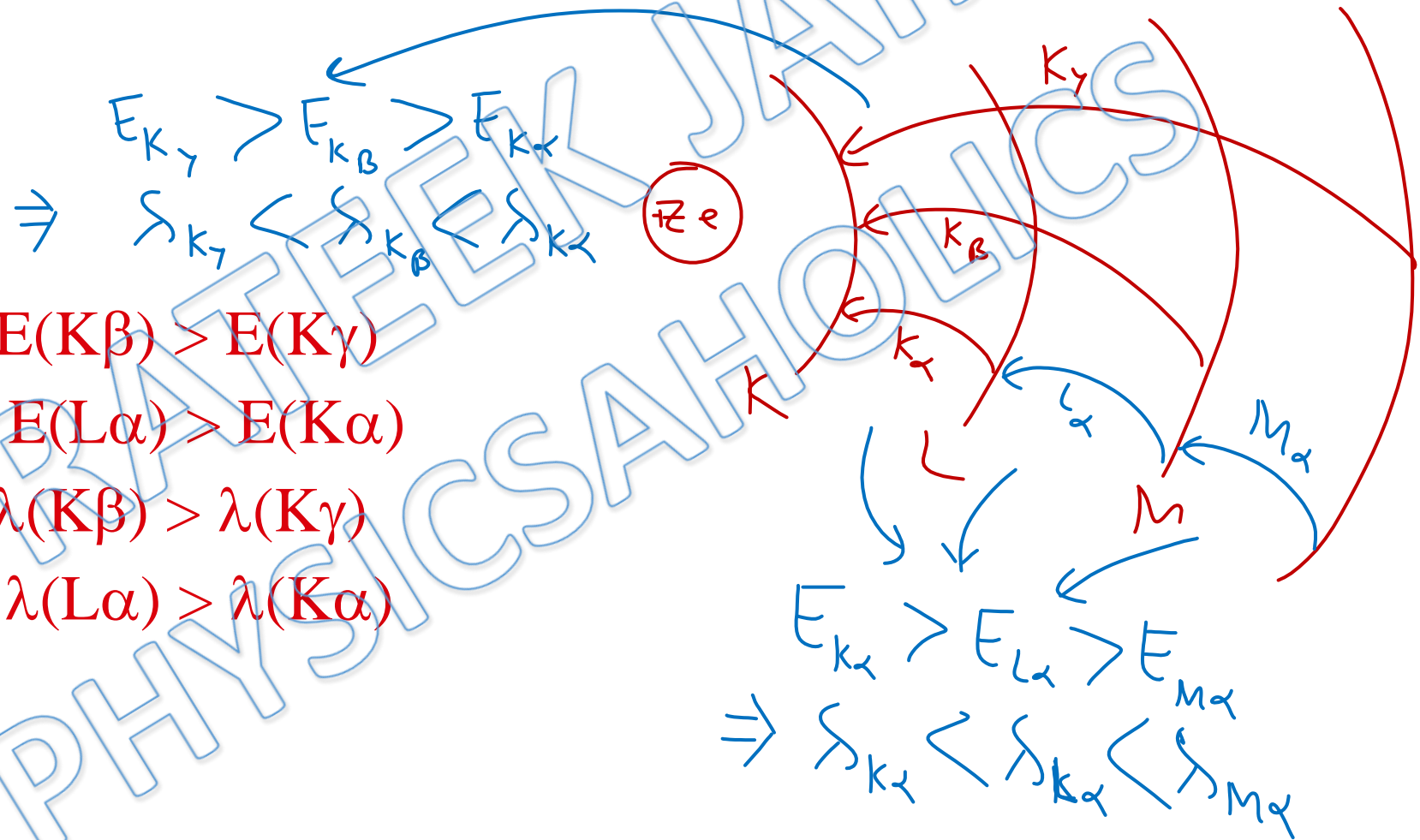
(d) 71 KeV

Maximum KE of photoelectron

$$= (124 - 82.7) \text{ KeV}$$

$$= 41 \text{ KeV}$$

Q2) For a given material, the energy and wavelength of characteristic x-ray satisfy—



- ~~(a)  $E(K\alpha) > E(K\beta) > E(K\gamma)$~~
- ~~(b)  $E(M\alpha) > E(L\alpha) > E(K\alpha)$~~
- ~~(c)  $\lambda(K\alpha) > \lambda(K\beta) > \lambda(K\gamma)$~~
- ~~(d)  $\lambda(M\alpha) > \lambda(L\alpha) > \lambda(K\alpha)$~~

## COMPREHENSION (Q3 to Q5)

A tungsten target ( $z = 74$ ) is bombarded by electrons in an X-ray tube. The K, L and M atomic X-rays energy levels for tungsten are 69.5, 11.3 and 2.3 keV

**Q3.** The minimum value of the accelerating potential that will permit the production of the characteristic  $k\beta$  and  $k\alpha$  lines of tungsten—

(a) 69.5 kV

(b) 58.2 kV

(c) 67.2 kV

(d) 11.3 kV

To remove K shell electron energy required

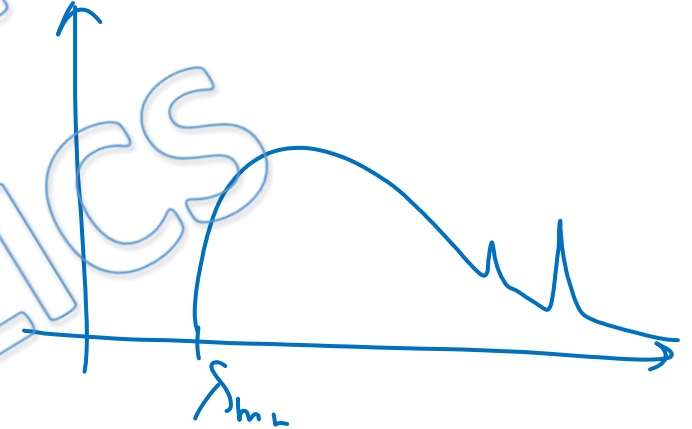
= 69.5 KeV

Required accelerating potential = 69.5 kV



Q4) For the same accelerating potential, what is  $\lambda_{min}$ ?

$$\begin{aligned}\lambda_{min} &= \frac{hc}{eV} = \frac{12400 \text{ A eV}}{69.5 \text{ KeV}} \\ &= \frac{1240000 \text{ Pm}}{69.5 \times 10^3} \\ &= 17.9 \text{ Pm}\end{aligned}$$



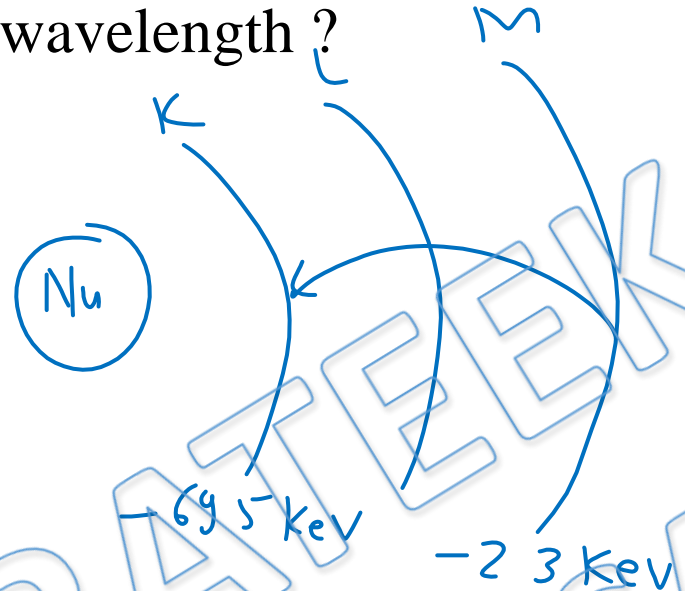
(a) 16.9 pm

(b) 17.9 pm

(c) 18.9 pm

(d) 19.9 pm

Q5) What is  $k\beta$  wavelength ?



$$\text{energy of photon} = (69.5 - 2.3) \text{ KeV} \\ = 67.2 \text{ KeV}$$

$$\lambda = \frac{hc}{E} \\ = \frac{1240000 \text{ pm eV}}{67.2 \text{ KeV}}$$

(a) 16.5 pm

(b) 17.5 pm

(c) 18.5 pm

(d) 21.5 pm

Q6)  $K_\alpha$  wavelength emitted by an atom of atomic number  $Z = 11$  is  $\lambda$ . Find the atomic number for an atom that emits  $K_\alpha$  radiation with wavelength  $4\lambda$  :

$$\sqrt{\nu} = a(z-b) \quad \text{for } K_\alpha, b=1$$

$$\nu = a^2(z-b)^2$$

(a)  $Z = 6$

(b)  $Z = 4$

(c)  $Z = 11$

(d)  $Z = 44$

$$\frac{c}{\lambda} = a^2(11-1)^2$$

$$\frac{c}{4\lambda} = a^2(z-1)^2$$

$$4 = \frac{100}{(z-1)^2}$$

$$(z-1)^2 = 25$$

$$z-1 = 5$$

$$\underline{z = 6}$$

Q7) X-rays will not show the phenomenon of :

(a) diffraction

(b) polarisation

(c) deflection by electric field

(d) interference

Q8) The wavelength of  $K_{\alpha}$  X-rays produced by an X-ray tube is  $0.76 \text{ \AA}$ . The atomic number of the anticathode material is

$$\sqrt{\nu} = a(z-b) \quad \text{for } K_{\alpha}, \quad a = \sqrt{\frac{3Rc}{4}}, \quad b=1$$

$$\nu = a^2 (z-b)^2$$

$$\Rightarrow \frac{c}{\lambda} = \frac{3Rc}{4} (z-1)^2$$

$$\begin{array}{r} .76 \times 33 \\ \hline 228 \\ 2286 \\ \hline 2508 \end{array}$$

(a) 82

~~(b) 41~~

(c) 20

(d) 10

$$\begin{aligned} (z-1)^2 &= \frac{4}{3 \times 76 \times 1.1 \times 10^7 \times 10^{-10}} = \frac{4000}{.76 \times 33} \\ &= \frac{4000}{2.5} = \frac{40000}{25} \end{aligned}$$

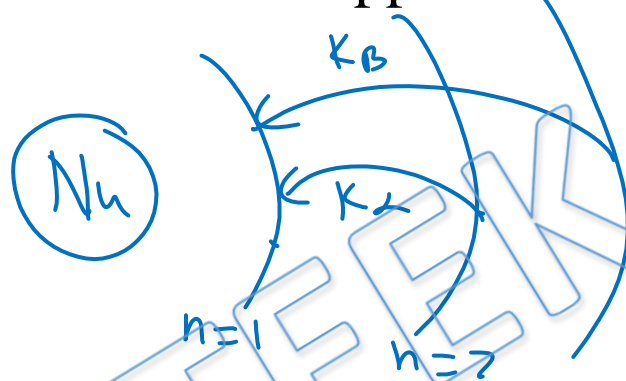
$$z-1 = \frac{200}{5} = 40 \Rightarrow z = 41$$

Q9) The minimum wavelength of X-ray that can be produced in a Coolidge tube depends on

$$\lambda_{\min} = \frac{hc}{eV}$$

- (a) the metal used as the target
- (b) the intensity of the electron beam striking the target
- (c) the current flowing through the filament
- (d) the potential difference between the cathode and the anode

Q10) If the potential difference applied across a Coolidge tube is increased,



- (a) the wavelength of the  $K_\alpha$  line will increase
- (b) the wavelength of the  $K_\beta$  line will decrease
- (c) the difference in wavelength between the  $K_\alpha$  and  $K_\beta$  lines will decrease
- (d) none of the above

Q11) If the voltage applied to a X-ray tube is being increased  $\eta = 1.5$  times, the short wave limit of the X-ray continuous spectrum shifts by  $\Delta\lambda = 26$  pm. Find the initial voltage applied to the tube.

- (a) 10 KV
- (b) 12 KV
- (c) 14 KV
- (d) 16 KV

$$\lambda_{\min} = \lambda = \frac{hc}{eV}$$

$$\lambda - \Delta\lambda = \frac{hc}{e \times 1.5V} = \frac{2hc}{3eV}$$

$$\frac{hc}{eV} - \Delta\lambda = \frac{2hc}{3eV}$$

$$\frac{hc}{3eV} = \Delta\lambda = 26 \text{ pm}$$

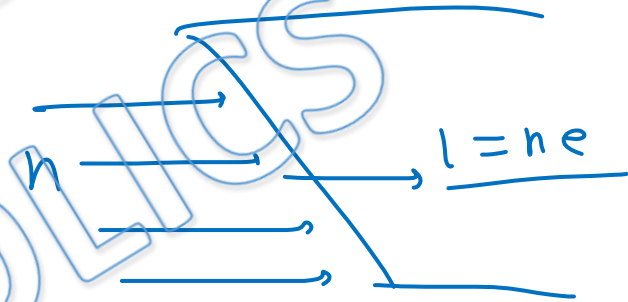
$$V = \frac{hc}{3e \times 26 \text{ pm}} = \frac{1240000 \text{ pm eV}}{3 \times 26 \text{ e pm}} = \frac{1240}{78} \text{ KV} = 16 \text{ KV}$$

$$\begin{array}{r} 31 \\ +240 \\ \hline 280 \\ = 155 \end{array}$$



Q12) The potential difference applied to an X-ray tube is 5 kV and the current through it is 3.2 mA. Then the number of electrons striking the target per second is

$$n = \frac{i}{e} = \frac{3.2 \times 10^{-3}}{1.6 \times 10^{-19}} = 2 \times 10^{16}$$



~~(a)  $2 \times 10^{16}$~~

(b)  $5 \times 10^6$

(c)  $1 \times 10^{17}$

(d)  $4 \times 10^{15}$

Q13) When the voltage applied to an X-ray tube is increased from  $V_1 = 10$  kV to  $V_2 = 20$  kV, the wavelength interval between the  $K_\alpha$  line and the short-wave cut off of the continuous X-ray spectrum increases by a factor  $\eta = 3.0$ . Find the atomic number of the element of which the tube's anticathode is made.

$$\text{Initial } \lambda_{\min} = \frac{hc}{eV} = \frac{12400 \text{ A eV}}{10 \text{ keV}} = 1240 \text{ \AA}$$

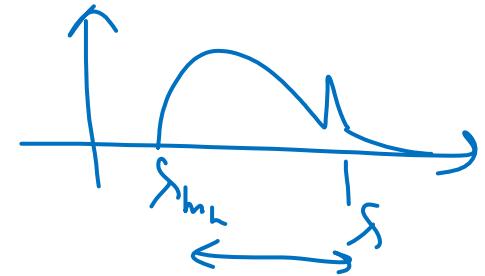
$$\text{final } \lambda_{\min} = \frac{12400 \text{ A eV}}{20 \text{ keV}} = \frac{1240}{2} = 620 \text{ \AA}$$

↑ wavelength of  $K_\alpha$

$$\lambda - 620 = 3(\lambda - 1240)$$

$$\Rightarrow \lambda - 620 = 3\lambda - 3720 \Rightarrow 2\lambda = 3080$$

$$\Rightarrow \lambda = 1540 \text{ \AA}$$



(a) 11

(b) 20

~~(c) 29~~

(d) 38

Ans. c

$$\sqrt{v} = a(z-b) \Rightarrow v = a^2(z-b)^2$$

$$\Rightarrow \frac{\epsilon}{155 \times 10^{-10}} = \frac{3 \times 11 \times 10^7 \epsilon}{4} (z-1)^2$$

$$\Rightarrow (z-1)^2 = \frac{4}{155 \times 3 \times 11 \times \underbrace{10^{-10} \times 10^7}} = \frac{4000}{155 \times 33}$$

$$(z-1)^2 = \frac{40000}{155 \times 33} = \frac{40000}{51.15}$$

$$\Rightarrow z-1 \approx \frac{200}{7} \approx 28$$

$$\Rightarrow \underline{\underline{z = 29}}$$

$$\begin{array}{r} 155 \times 33 \\ \hline 465 \\ 4650 \\ \hline 5115 \end{array}$$

Q14) If a potential difference of 20,000 volts is applied across an X-ray tube, the cut-off wavelength will be

$$\lambda_{\text{cutoff}} = \frac{hc}{eV} = \frac{12400 \text{ A eV}}{20000 \text{ eV}} = \frac{1240 \times 10^{-11}}{200} = 62 \times 10^{-11}$$

(a)  $6.21 \times 10^{-10} \text{ m}$

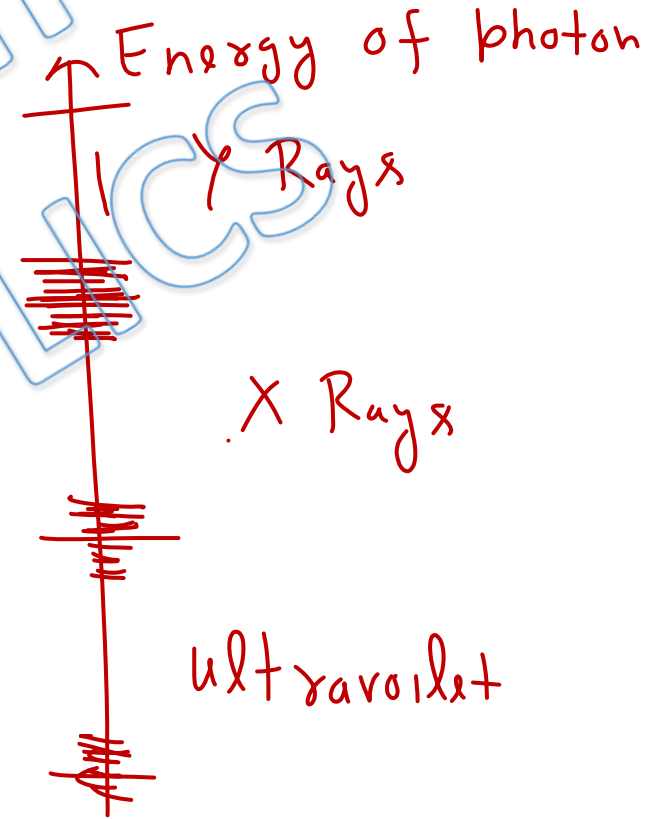
(b)  $6.21 \times 10^{-11} \text{ m}$

(c)  $6.21 \times 10^{-12} \text{ m}$

(d)  $3.1 \times 10^{-11} \text{ m}$

Q15) Which of the following pairs constitute very similar radiations?

- (a) Hard ultraviolet rays and soft X-rays
- (b) Soft ultraviolet rays and hard X-rays
- (c) Very hard X-rays and low-frequency  $\gamma$ -rays
- (d) Soft X-rays and  $\gamma$ -rays



Q16) When an electron moving at a high speed strikes a metal surface, which of the following are possible?

- (a) The entire energy of the electron may be converted into an X-ray photon.
- (b) Any fraction of the energy of the electron may be converted into an X-ray photon.
- (c) The entire energy of the electron may get converted to heat.
- (d) The electron may undergo elastic collision with the metal surface.

Q17) The wavelength of  $K_{\alpha}$  X-rays for lead isotopes  $Pb^{208}$ ,  $Pb^{206}$ ,  $Pb^{204}$  are  $\lambda_1$ ,  $\lambda_2$  and  $\lambda_3$  respectively. Then :

$$\sqrt{\nu} = a(z - b)$$

$$\sqrt{\frac{c}{\lambda}} = a(z - b)$$

~~(a)~~  $\lambda_1 = \lambda_2 = \lambda_3$

(c)  $\lambda_1 < \lambda_2 < \lambda_3$

(b)  $\lambda_1 > \lambda_2 > \lambda_3$

~~(d)~~  $\lambda_2 = \sqrt{\lambda_1 \lambda_3}$

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